

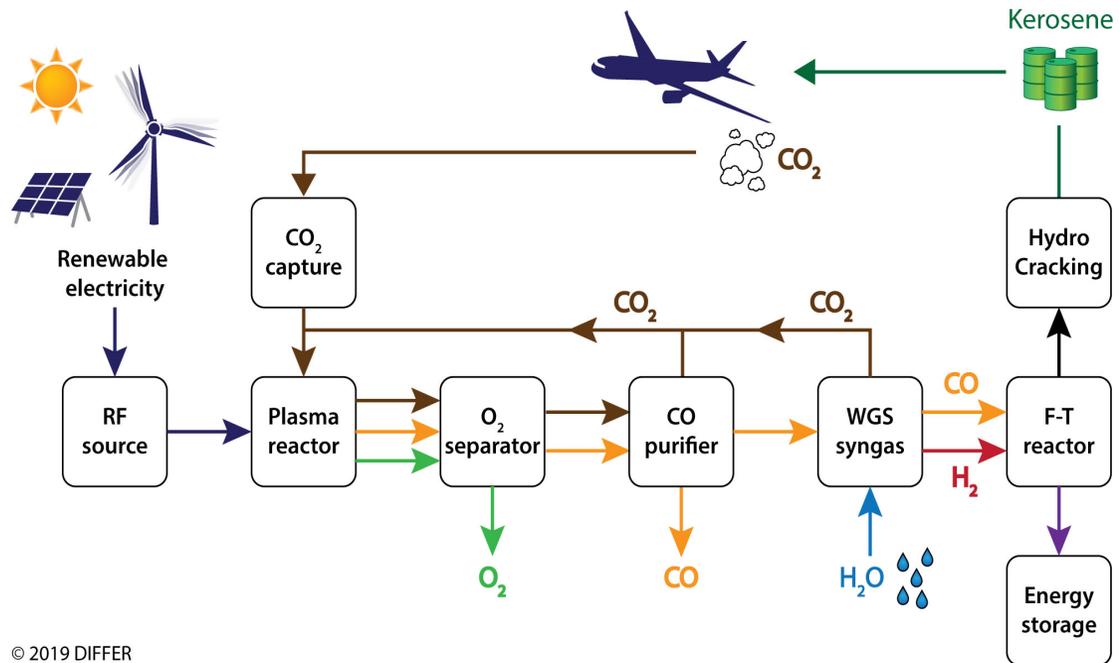


Production of Sustainable aircraft grade Kerosene from water and air powered by Renewable Electricity, through the splitting of CO₂, syngas formation and Fischer - Tropsch synthesis

The KEROGREEN CO₂ plasma route to CO and alternative fuels

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DUTCH I NSTITUTE F OR F UNDAMENTAL E NERGY R ESEARCH, EINDHOVEN, THE NETHERLANDS



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Kerogreen aim: Demonstration of the full chain process from renewable, electricity, CO₂ (captured) and H₂O to kerosene.

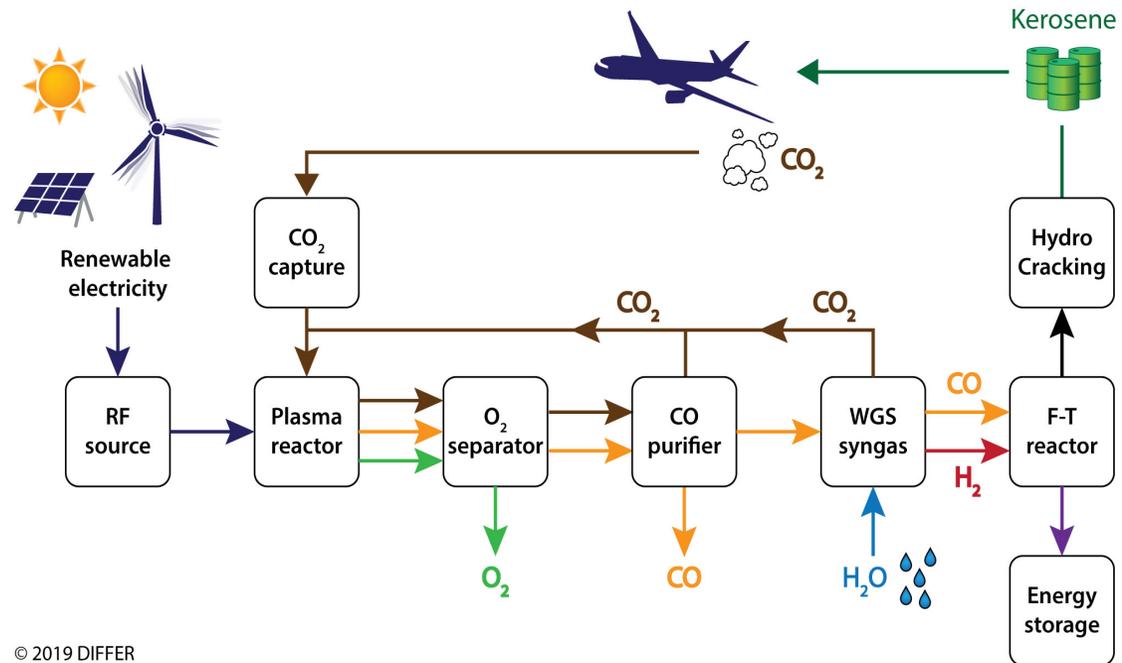
- Research and optimization of individual process steps **TRL (1-3) → 4**
- Integration phase at Karlsruhe Institute of Technology → **3 L per day**
- Duration 2018-2022

INERATEC



This project has received funding from the European Union's Horizon 2020 Research and Innovation Programme under GA-Nr. 763909





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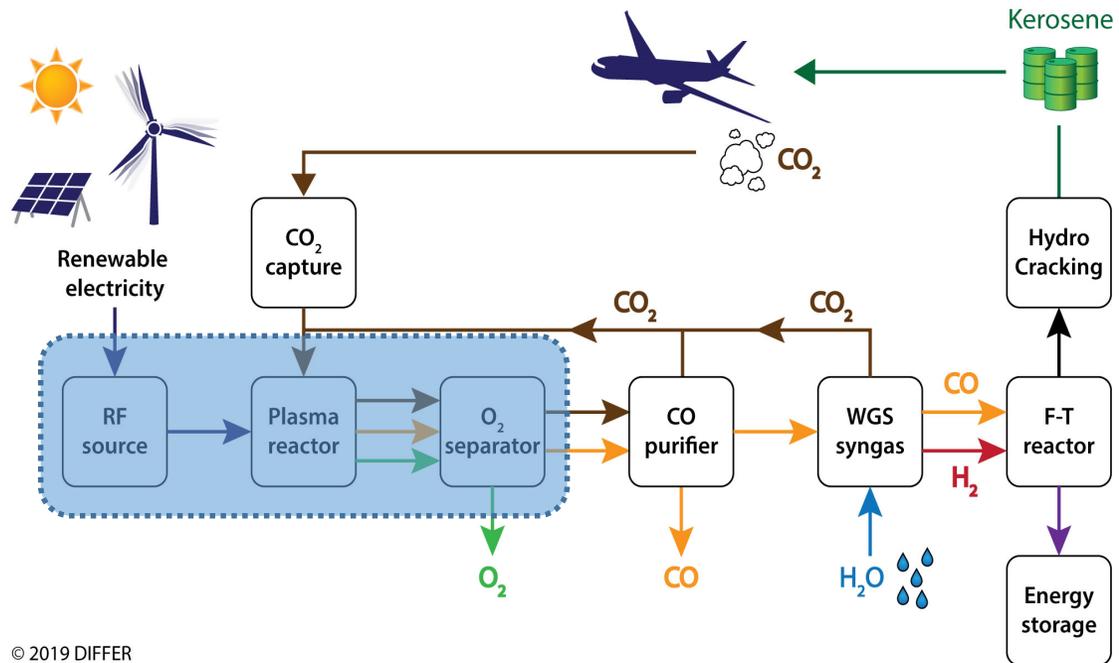
Main challenges

- Plasma dissociation efficiency of CO₂
- Oxygen separation after plasmolysis by SOEC
- System integration of different technologies into one container sized assembly
- Maximization of the energy and carbon efficiency of the full chain

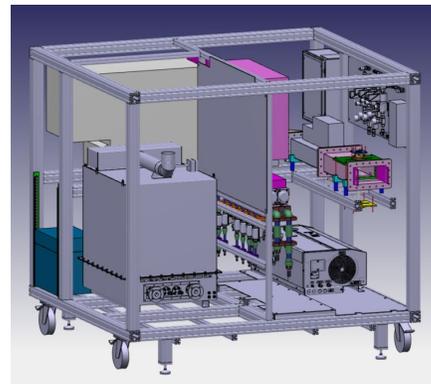
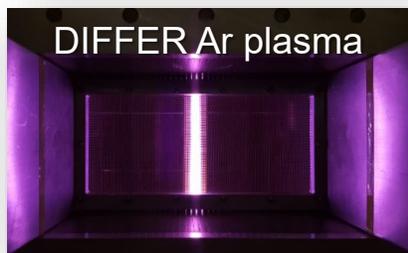
KEROGREEN offers an innovative conversion route based on **plasma driven dissociation of CO₂**, separation of oxygen by means of solid oxide electrolyte cells and Fischer-Tropsch (F-T) synthesis of kerosene.

- CO₂ plasmolysis (DIFFER)
- O₂ separation (DIFFER, VITO, Cerpotech, Hygear)
- CO purification (HYGEAR)
- Water gas shift reaction reaction (KIT)
- F-T synthesis (INERATEC)
- Heavy HC hydrocracking (KIY)



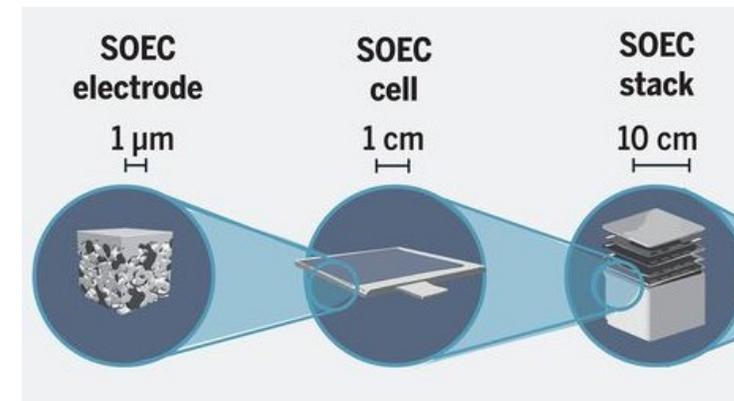


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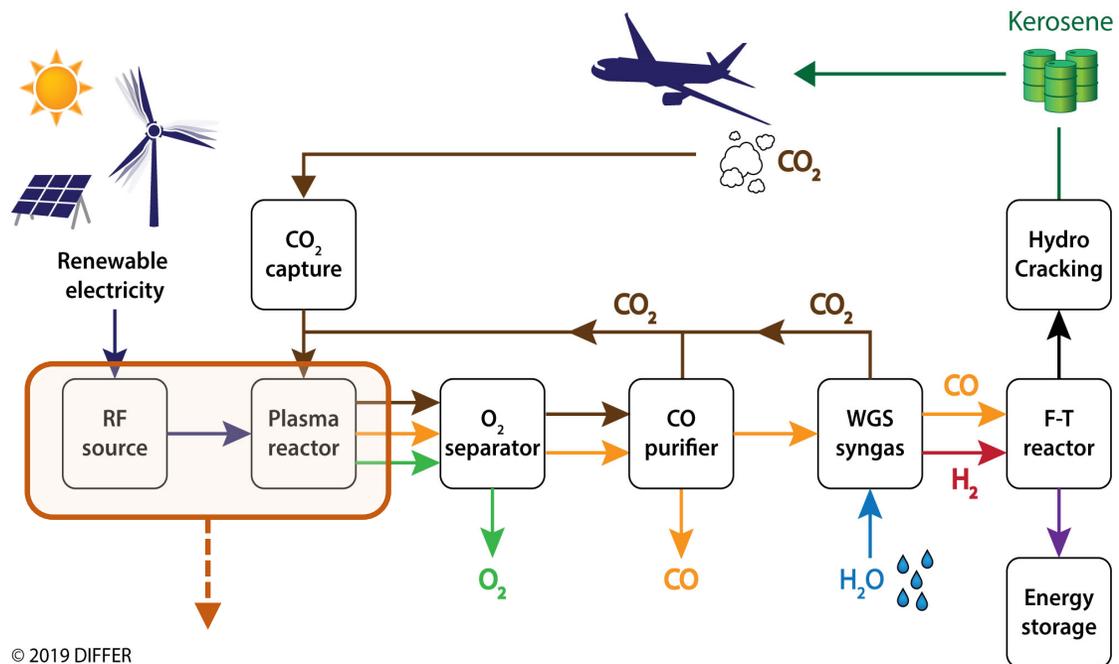
DIFFER involvement

- **Plasma modeling and optimization**
- Plasma upscaling from 1 to 6 kW
- Material requirements for using SOEC as oxygen separator
- SOEC upscaling from 1W to 1.5 kW



DOI: 10.1126/science.aba6118

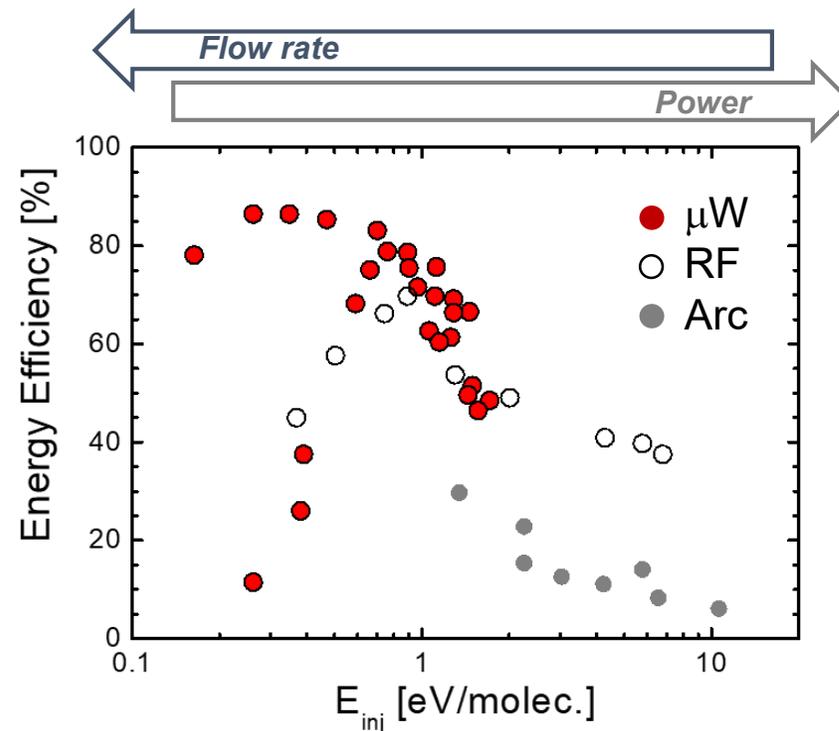
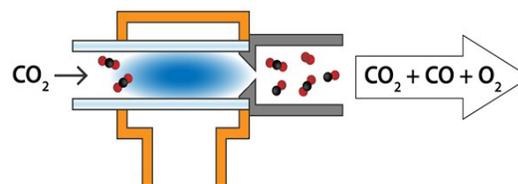
Why CO₂ plasmolysis?



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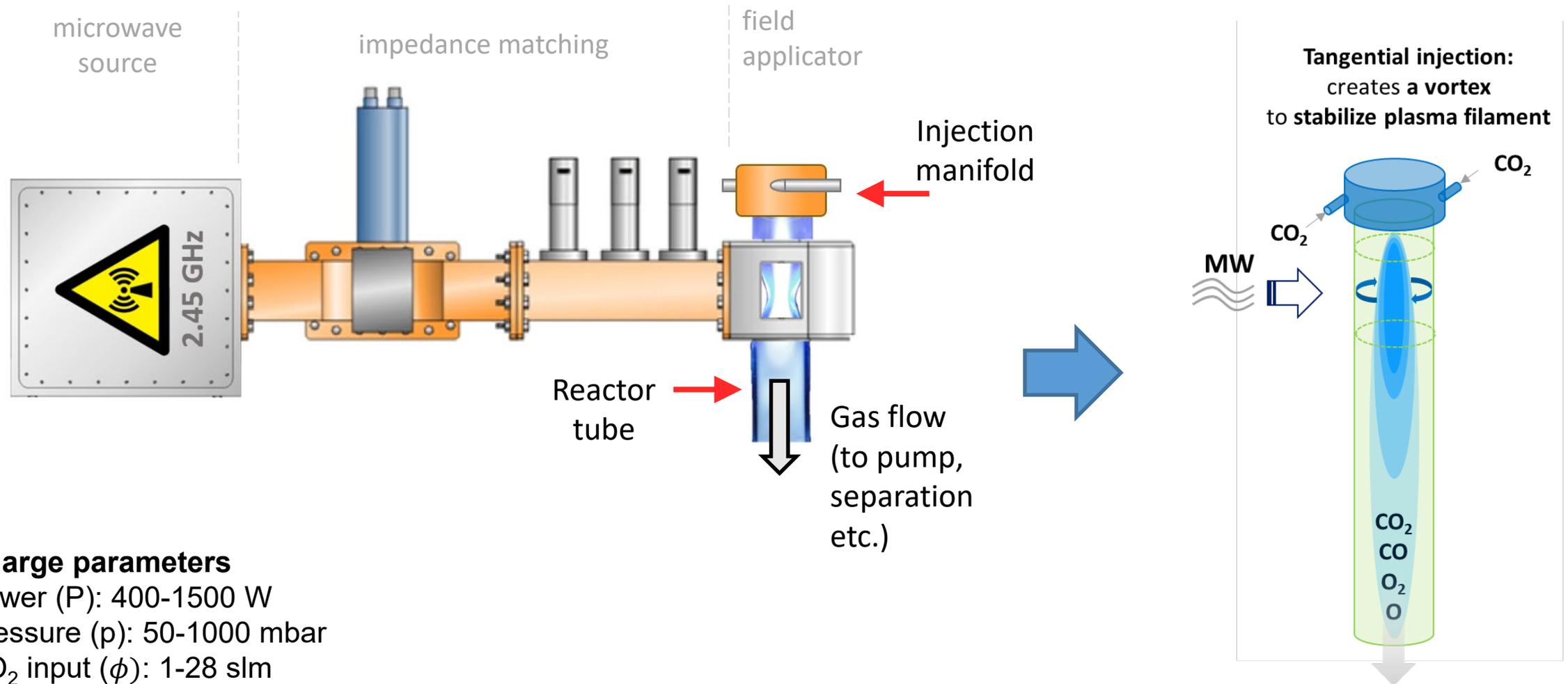
- Input: CO₂ + renewable electricity
- Output: CO₂, CO and O₂
- High energy efficiency, ...



DOI: 10.1017/CBO9780511546075

Microwave generated plasma so far most efficient

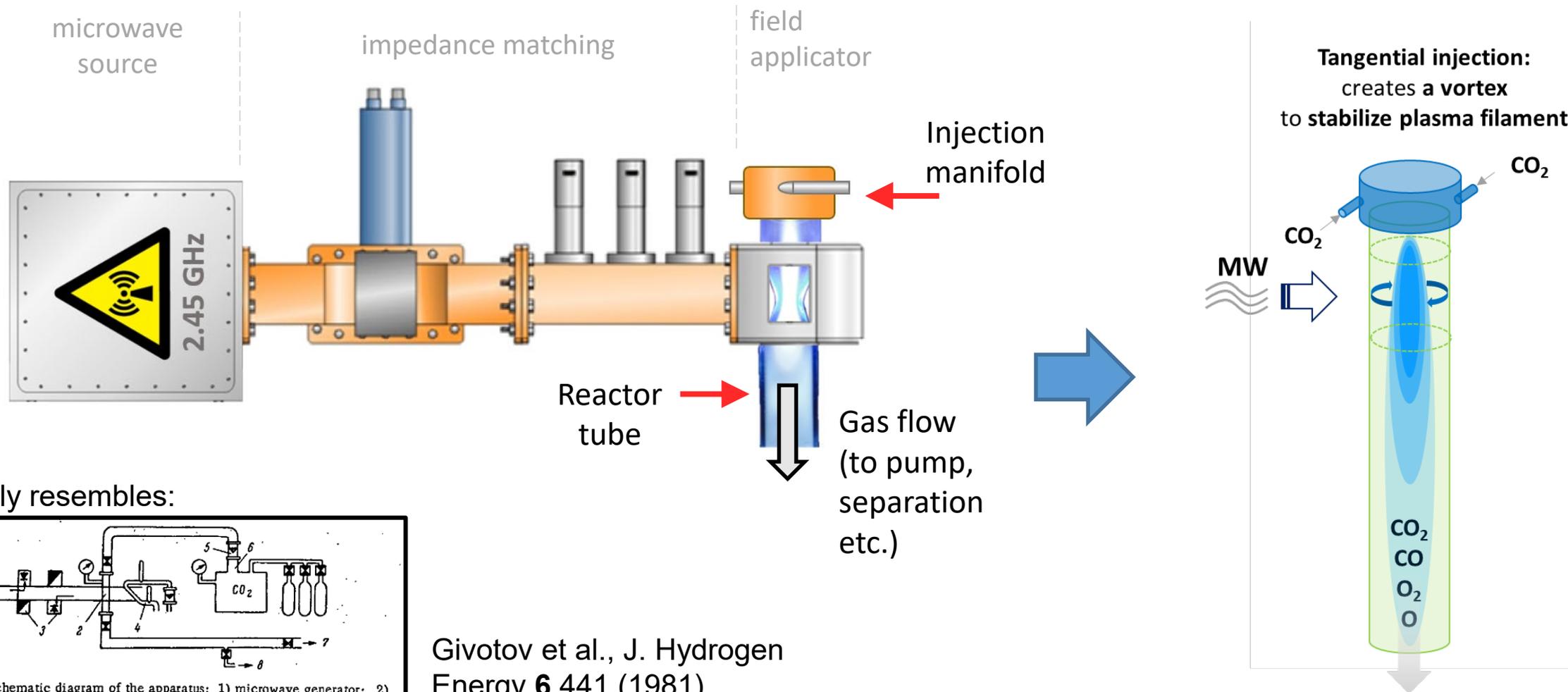
Set up - CO₂ conversion in a microwave plasma flow reactor



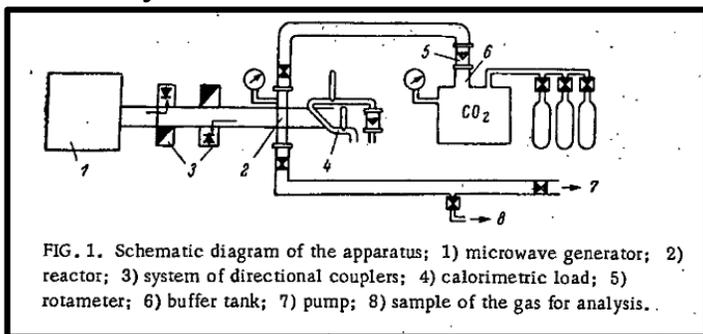
Discharge parameters

- Power (P): 400-1500 W
- Pressure (p): 50-1000 mbar
- CO₂ input (ϕ): 1-28 slm

Set up - CO₂ conversion in a microwave plasma flow reactor

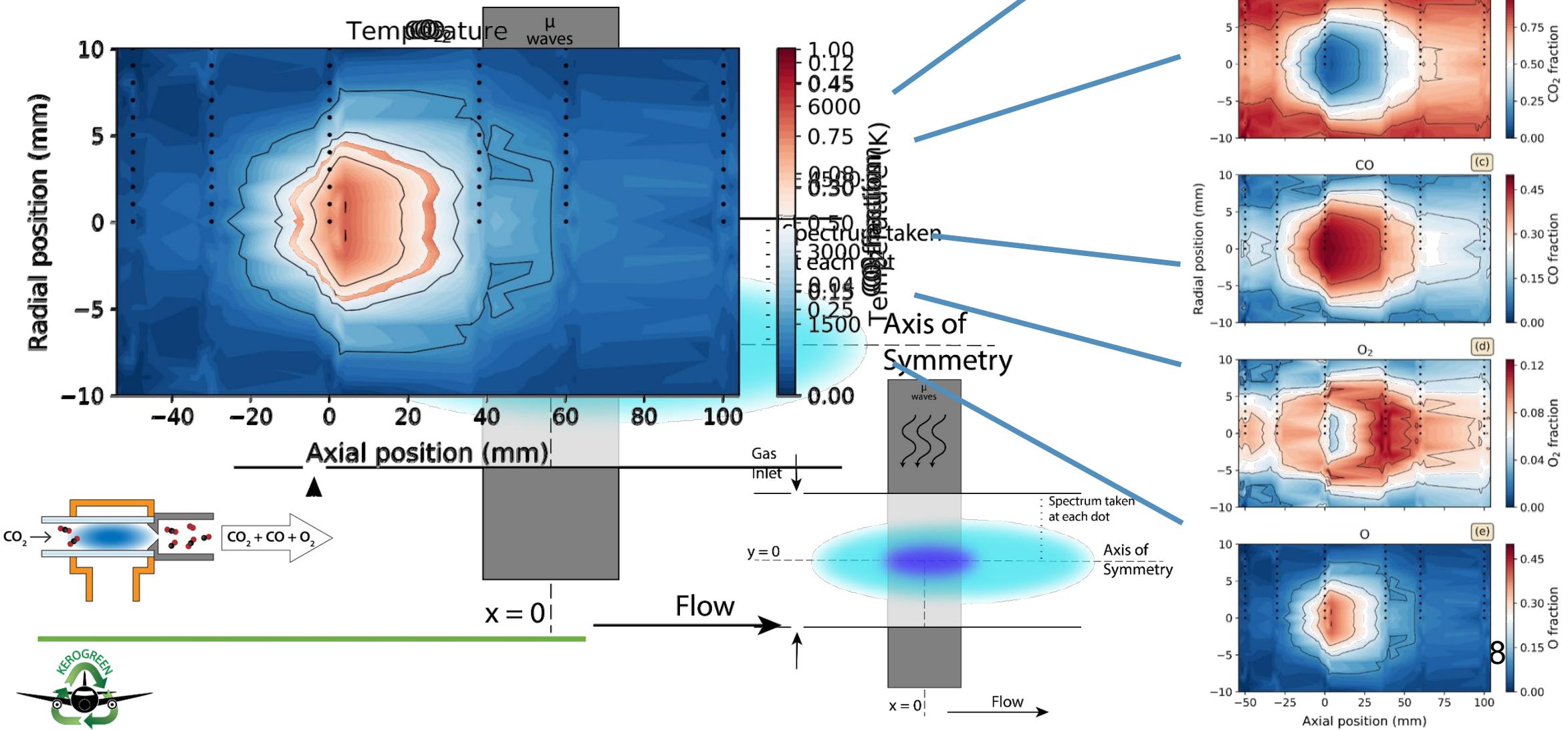


Closely resembles:

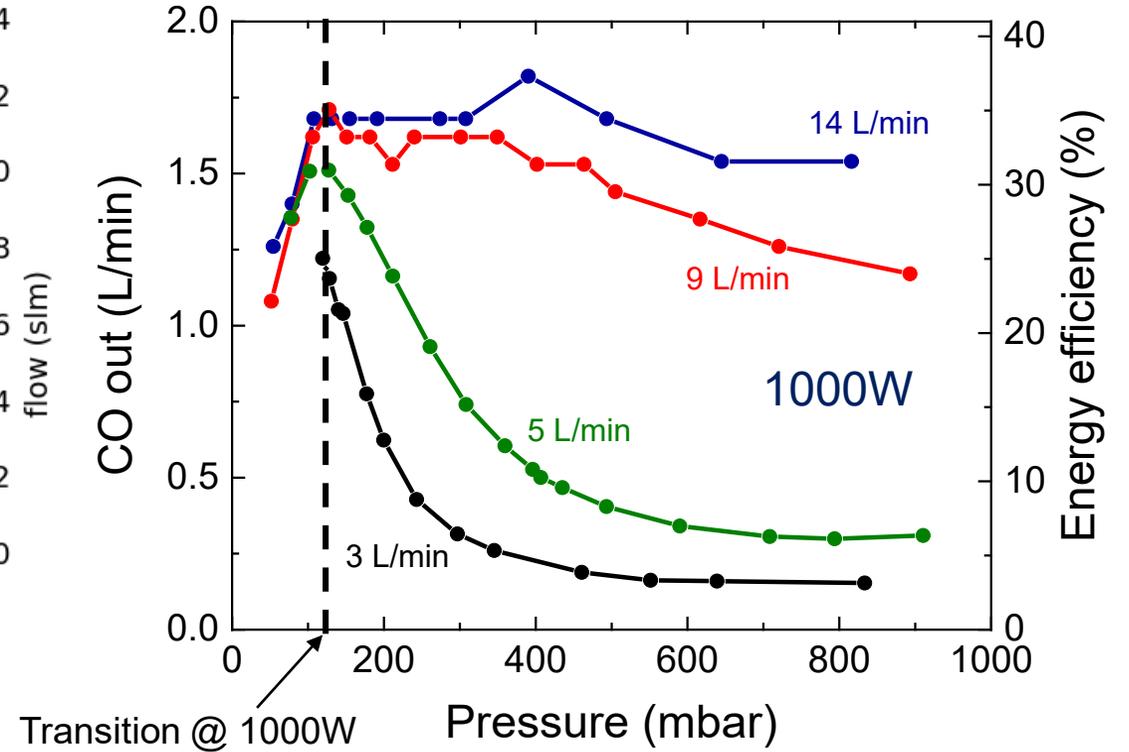
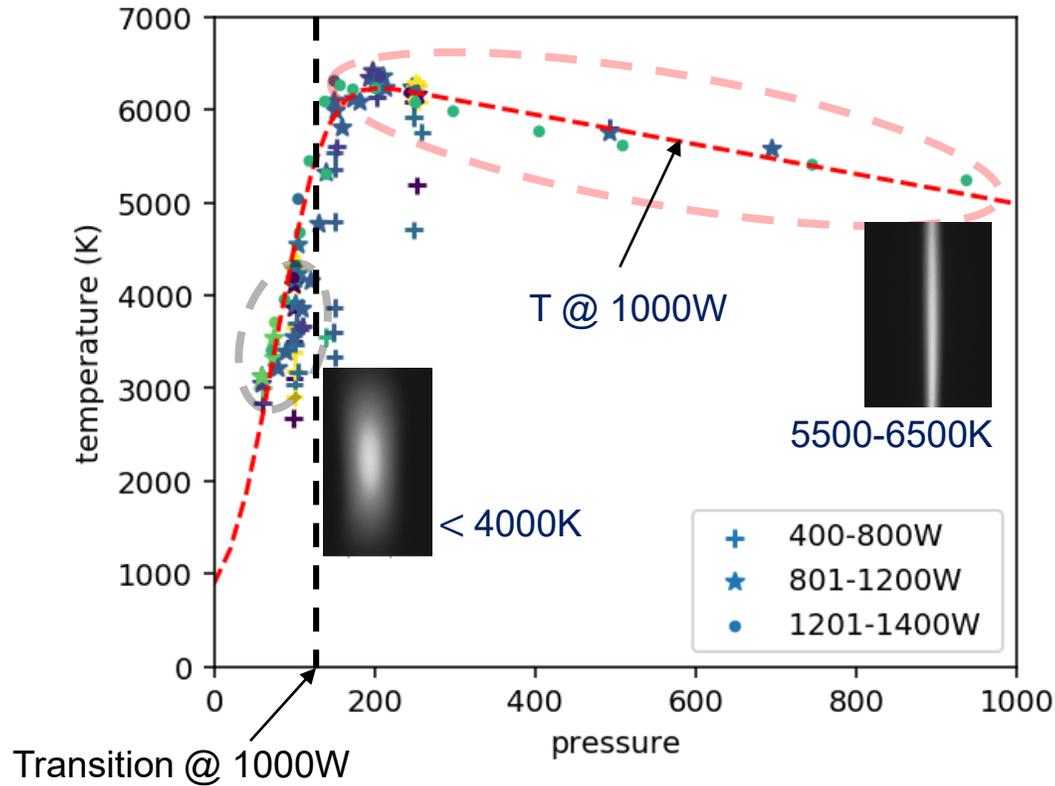


Givotov et al., J. Hydrogen Energy 6 441 (1981)
80% energy efficiency reported

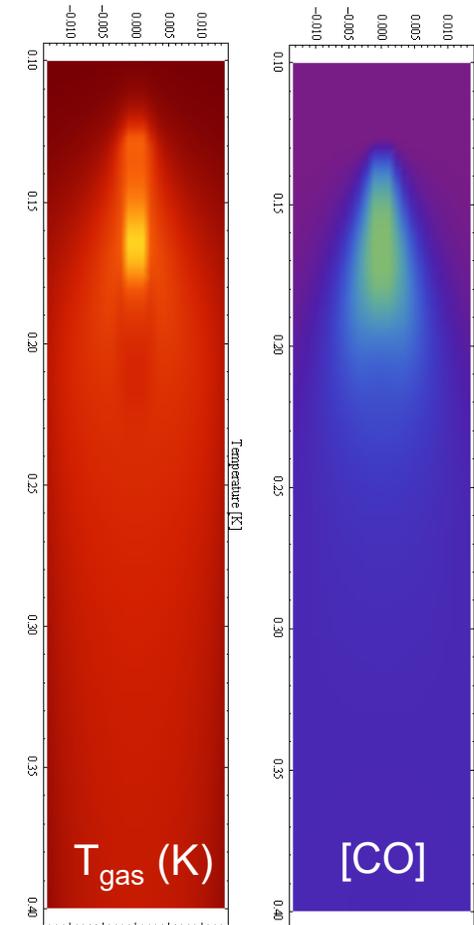
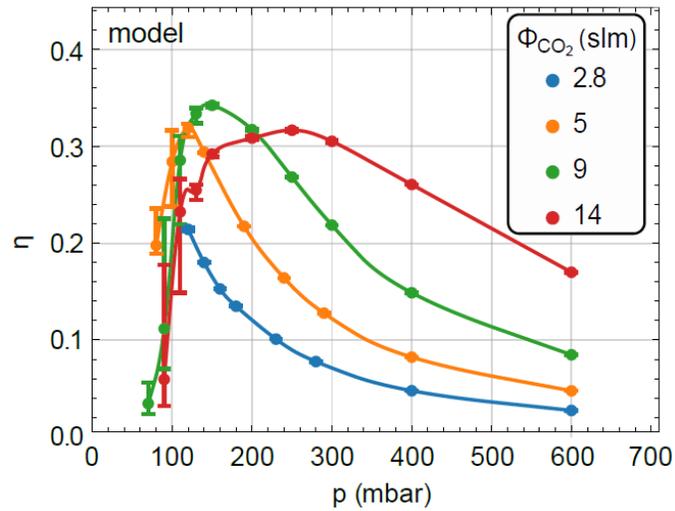
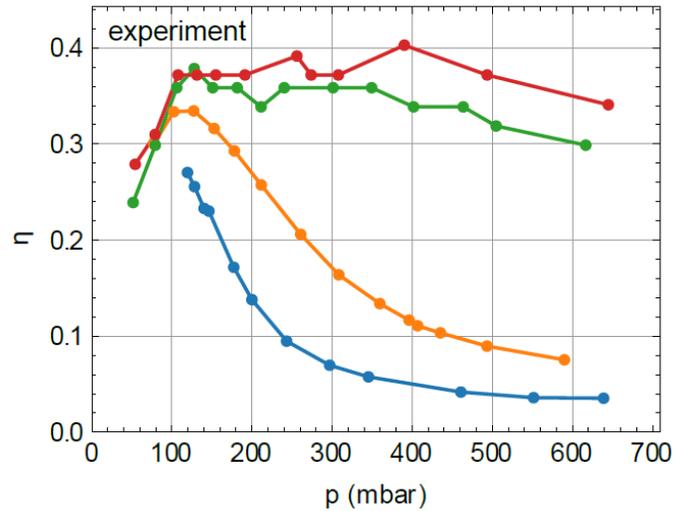
2D maps T & CO₂, CO, O₂ and O fractions



Gas temperature and CO output are correlated



T and plasma shape independent of flow rate...
 ... yet, strong flow dependence at high pressure → Transport effect!



- Both pressure and flow dependence of η roughly captured despite substantial simplifications in flow implementation
- Simple thermal chemistry adequate to describe reactor performance

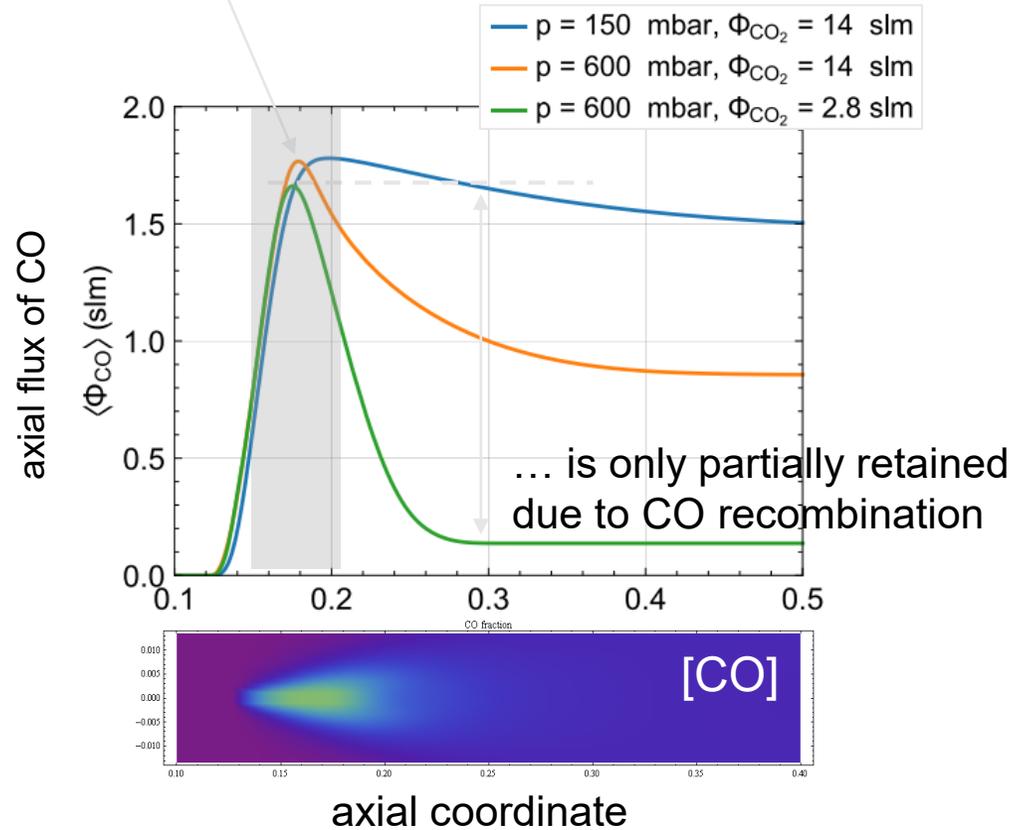
Wolf, A. J. and Peeters, F. J. J., et al. (2020) J. Phys. Chem. C 2020, 124, 31, 16806–16819



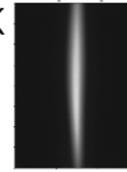
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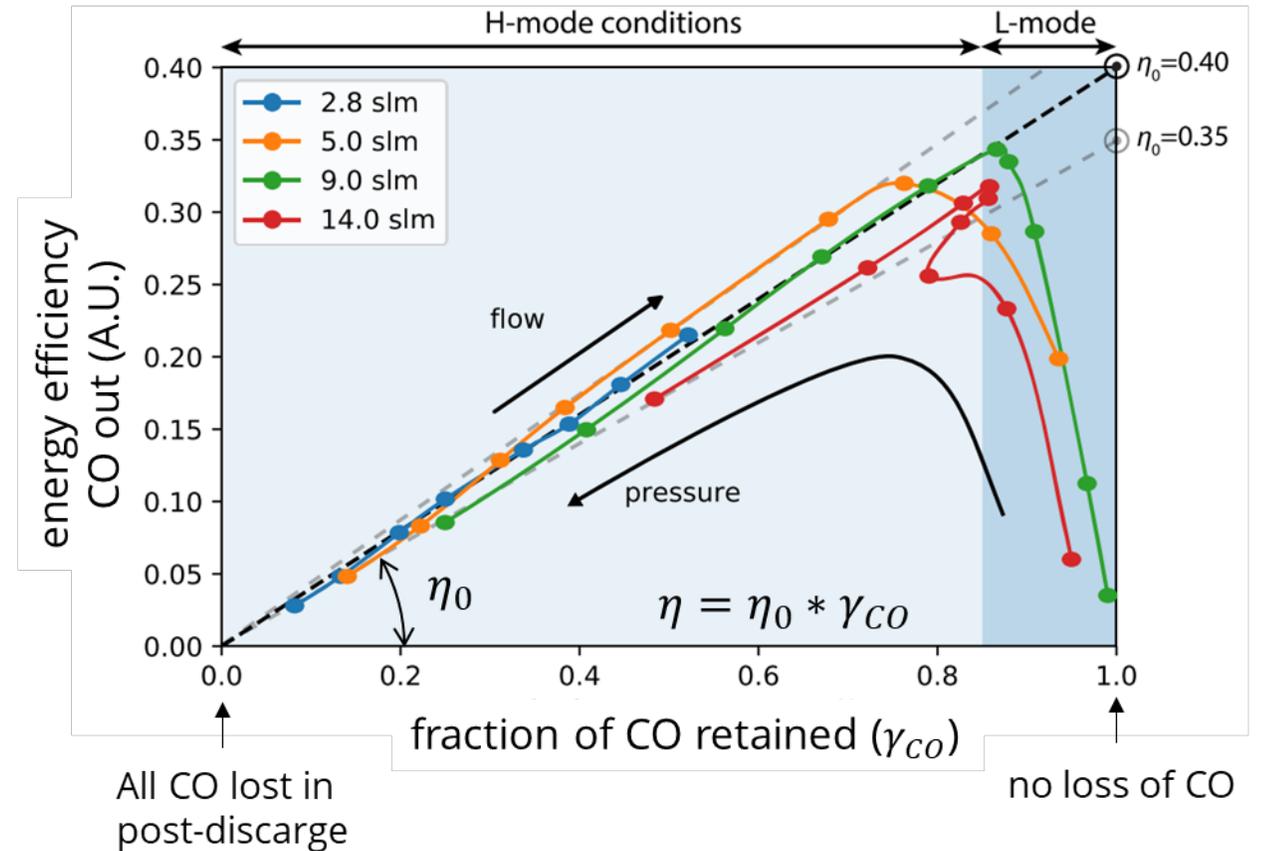
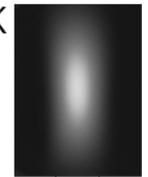
net CO flux generated in plasma region...

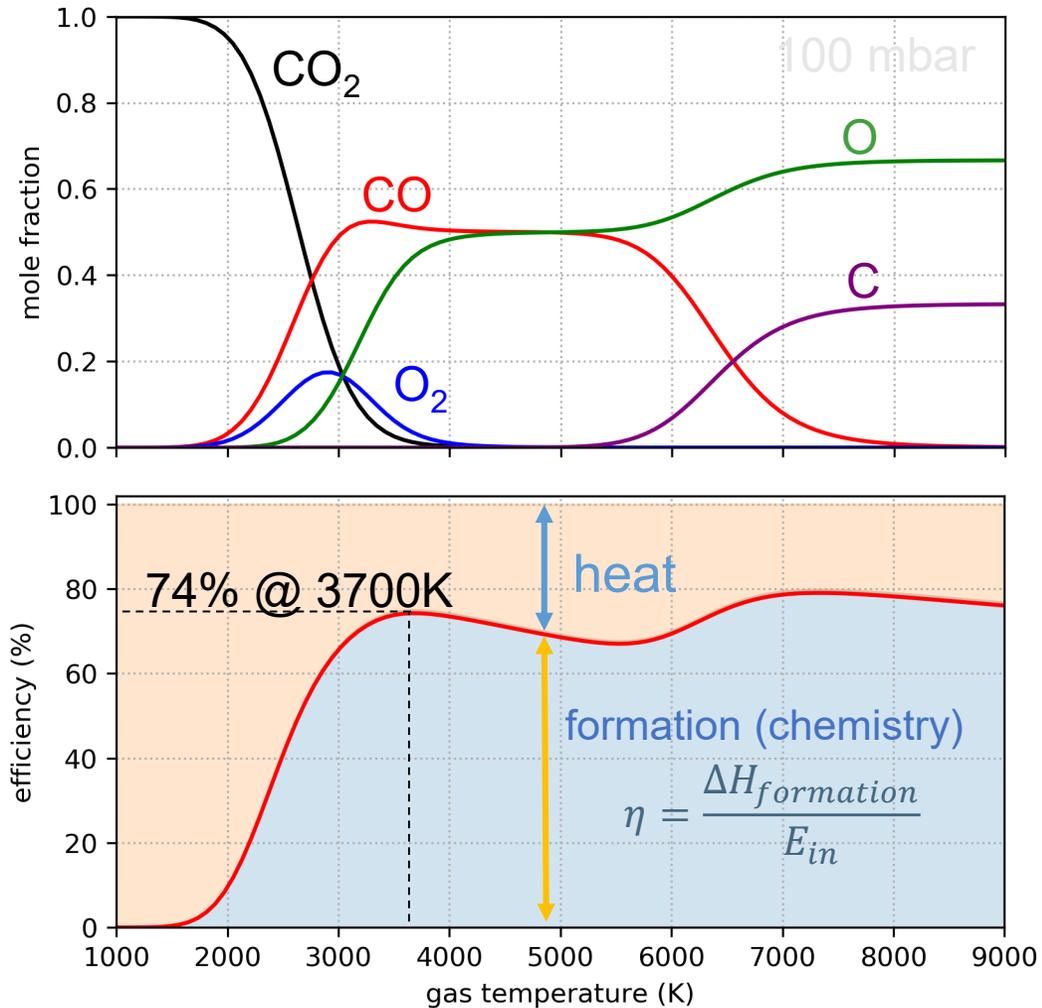


$T_{\text{gas}} \approx 6000\text{K}$



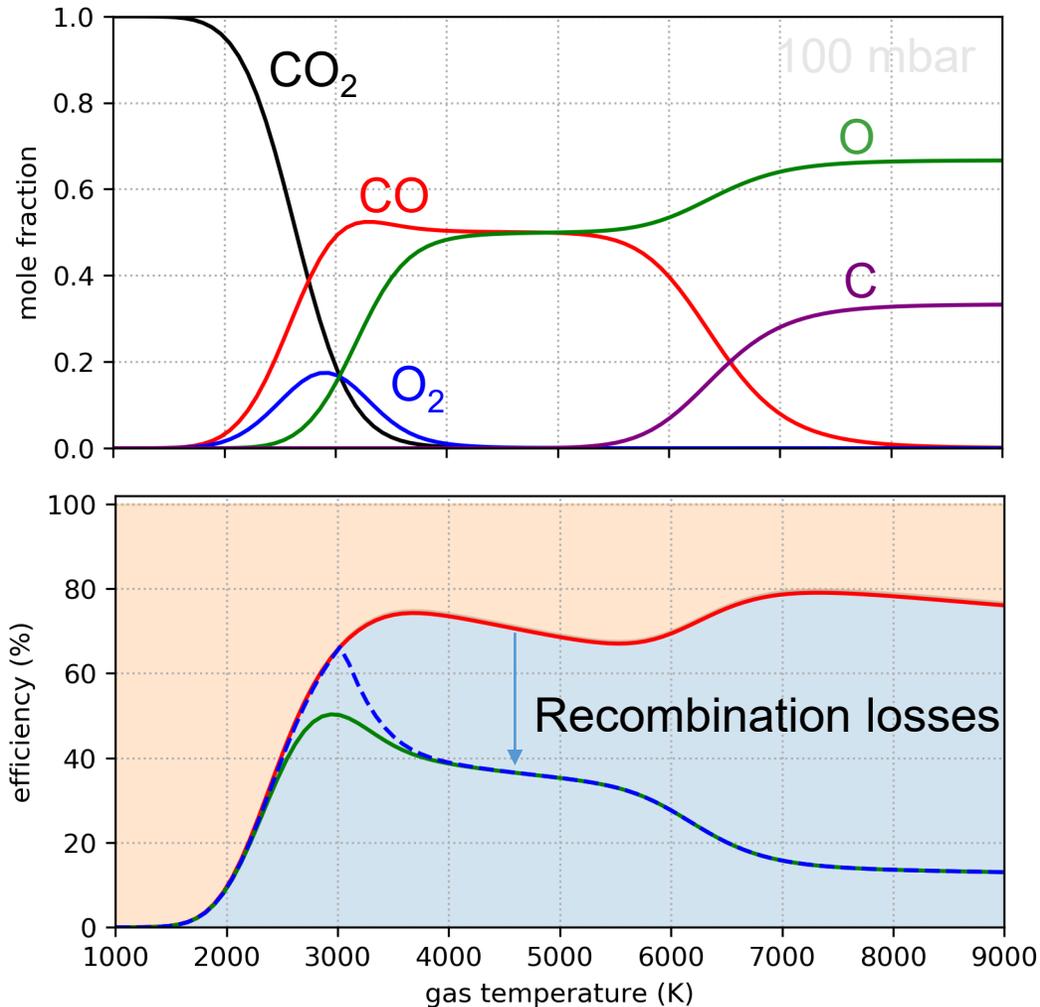
$T_{\text{gas}} < 4000\text{K}$





Thermodynamic equilibrium composition:
minimization of Gibbs free energy

The **fraction of energy invested in formation** depends on the **temperature**



3 quenching scenarios without CO back-reactions

"Ideal" quenching

- $O + O + (M) \rightarrow O_2 + (M)$
- 52% efficiency @ 3000 K

"Super-ideal" quenching

- $O + \text{CO}_2 \rightarrow \text{CO} + \text{O}_2$
- 65% efficiency @ 3000 K

Super-ideal quenching (open system)

- Admixture of extra CO_2
- > 70% efficiency



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Thank you for your attention!



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